

Title: APPARATUS AND METHOD FOR
REMOVING PHOSPHORUS FROM
WASTE LAGOON EFFLUENT
Applicant(s): Bowers et al.
Atty. Docket No.: 297/181

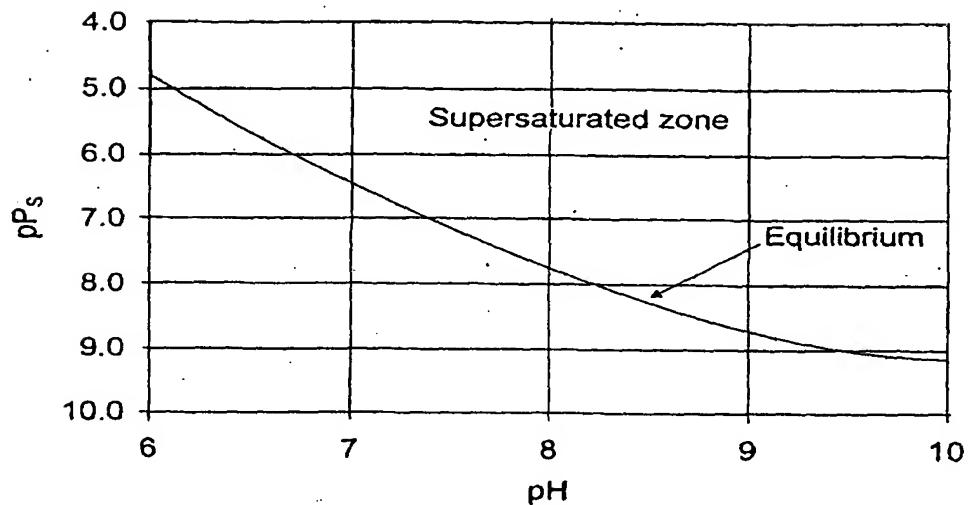


Figure 1: Variation of Equilibrium Conditional Solubility versus pH for Struvite
(from Ohlinger et al., 1998)

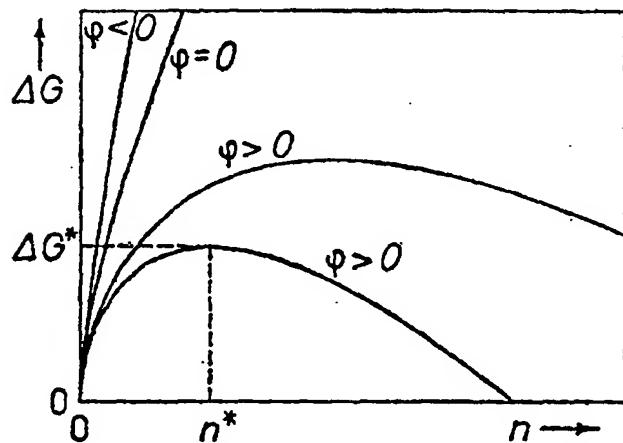


Figure 2: Free Energy versus Number of Particles in a Precipitating Crystal

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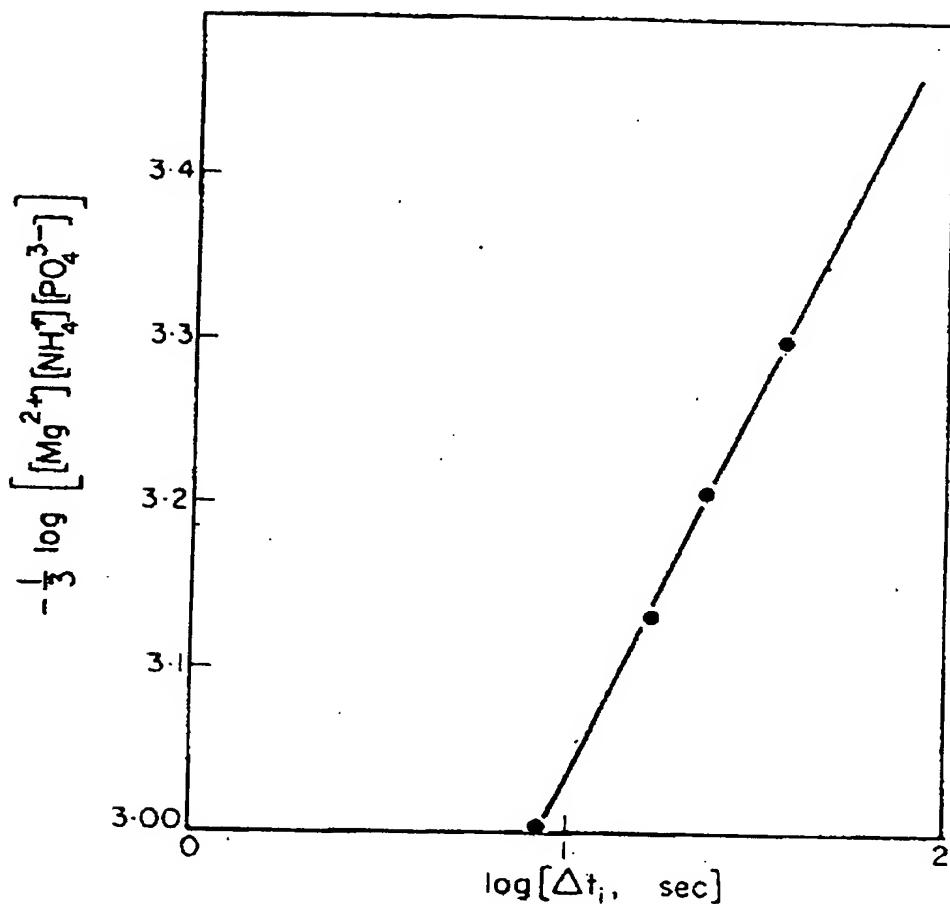


Figure 3: Concentration (-1/3 log of Ionic Product, Mol/L) versus Induction Time (Sec) for Struvite Precipitation

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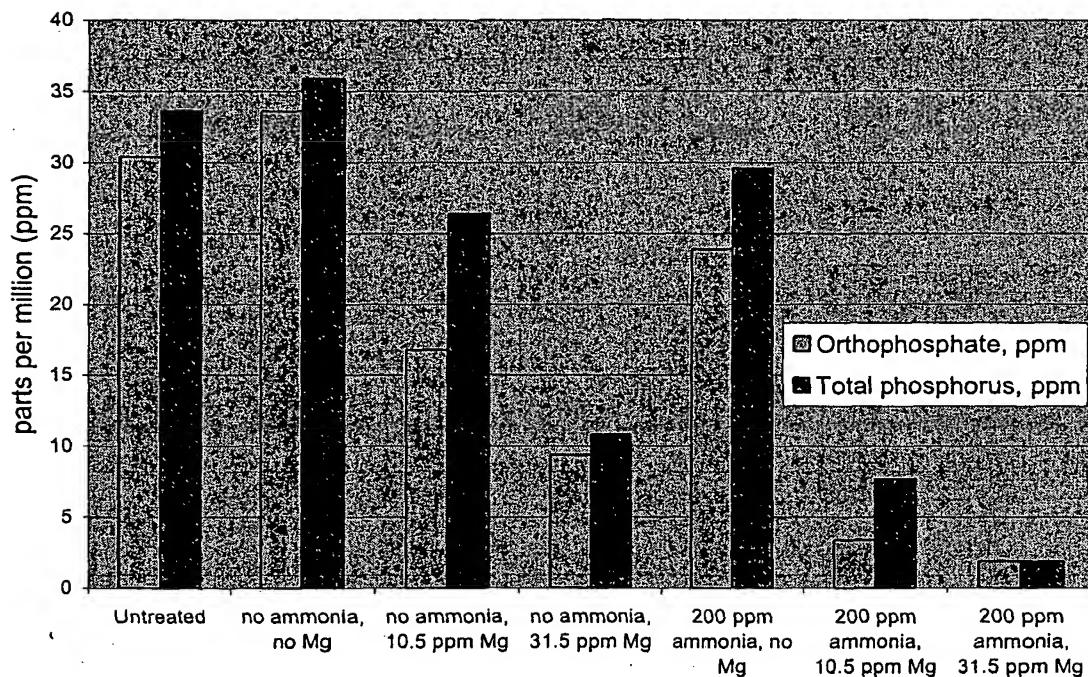


Figure 4: Dissolved OP and TP (ppm) in Untreated and Treated Effluent
from Rocky Mount Lagoon

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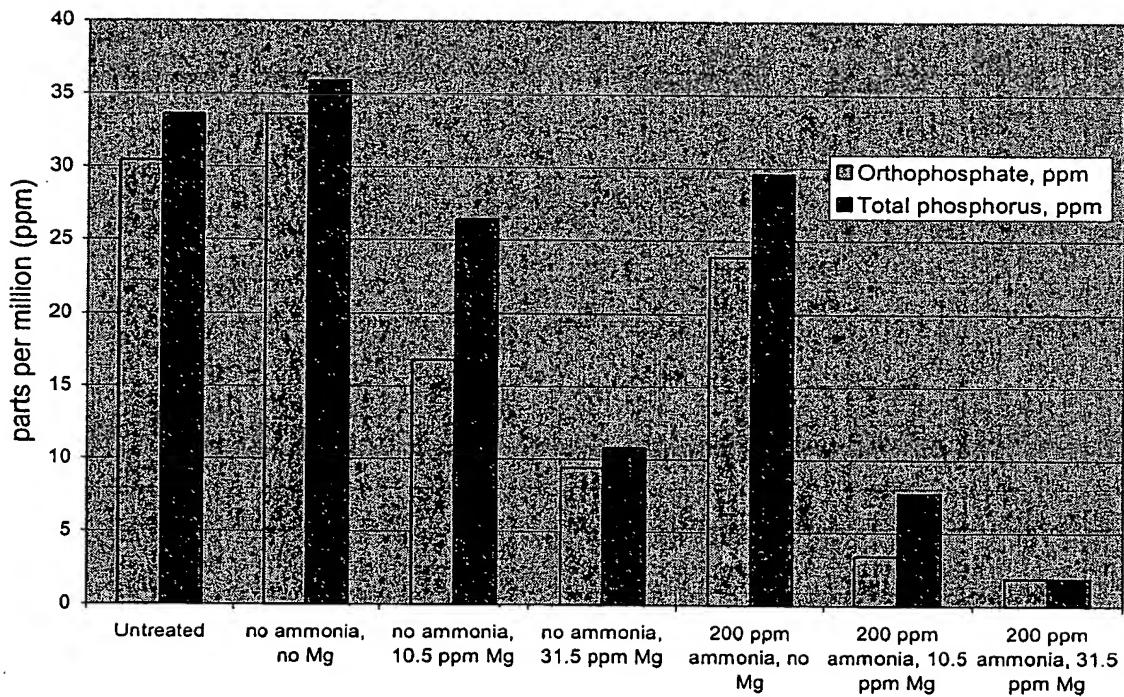
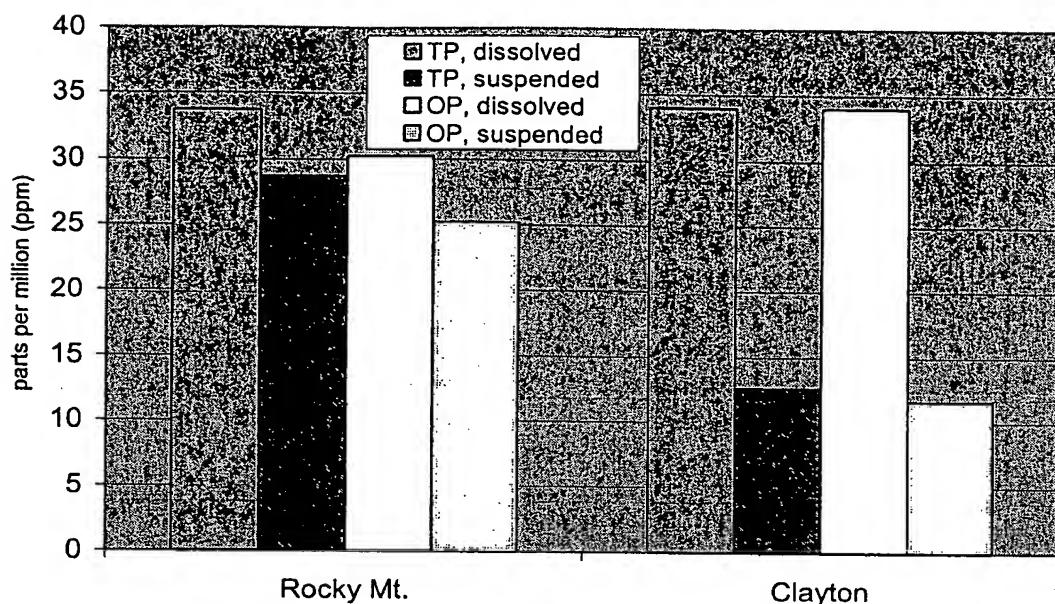
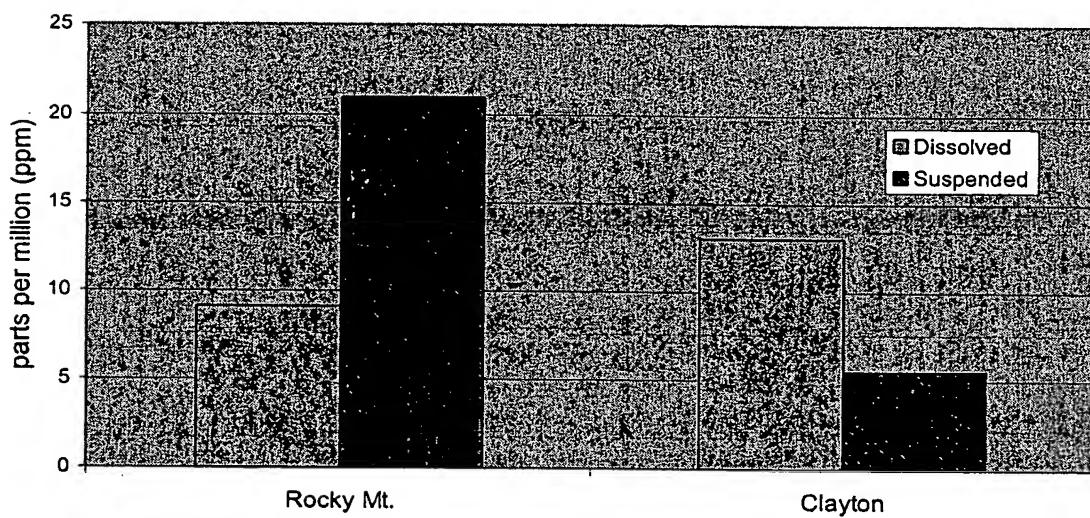


Figure 5: Dissolved OP and TP (ppm) in Untreated and Treated Effluent from Clayton Digester

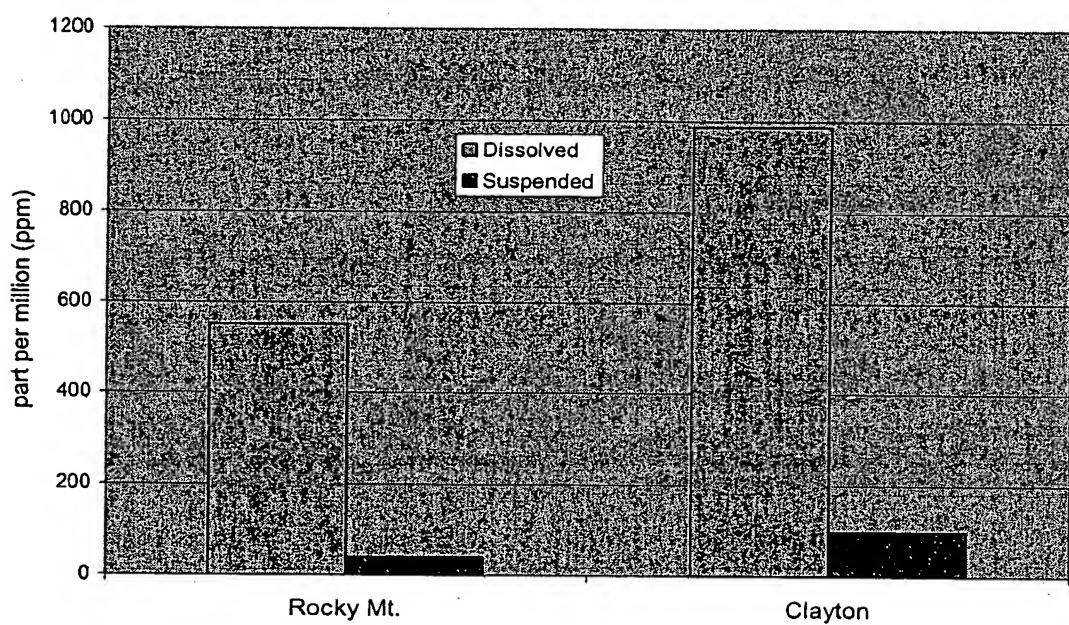


**Figure 6: Breakdown of Phosphorus Content (ppm) by Form in
Rocky Mount and Clayton Effluent**



**Figure 7: Breakdown of Mg Content (ppm) by Form in
Rocky Mount and Clayton Effluents**

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**Figure 8: Breakdown of TAN (ppm) by Form
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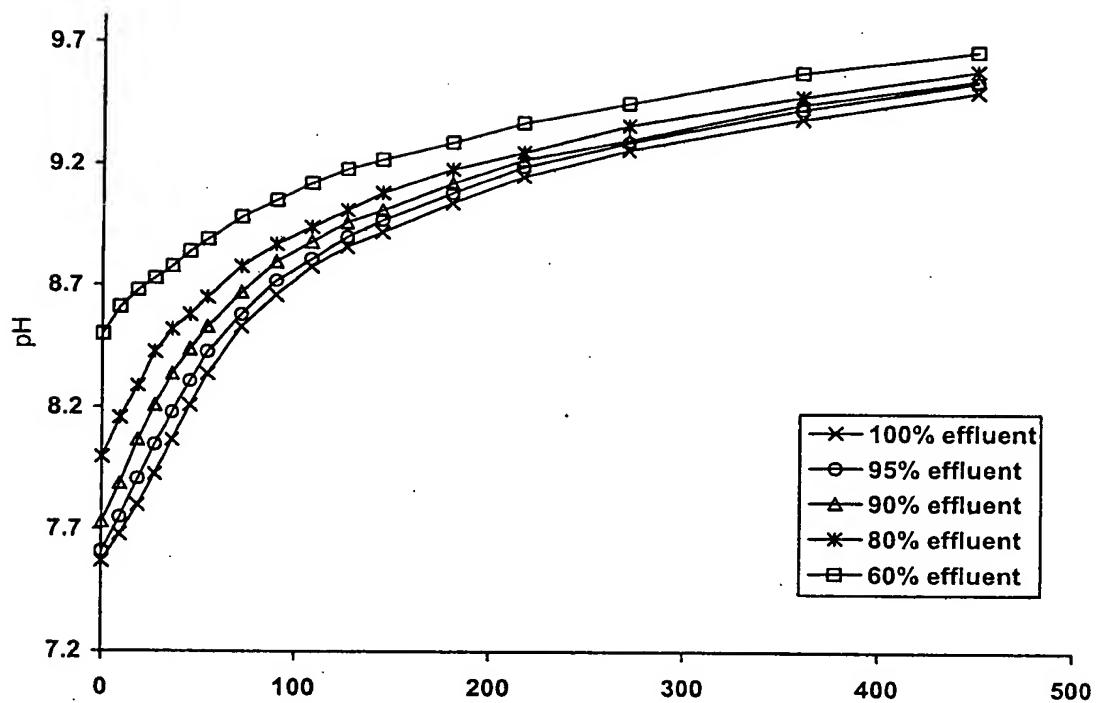


Figure 9: pH versus Amount of Ammonia Added (ppm) for Five Ratios of Effluent to Mg-Supplementing Solution

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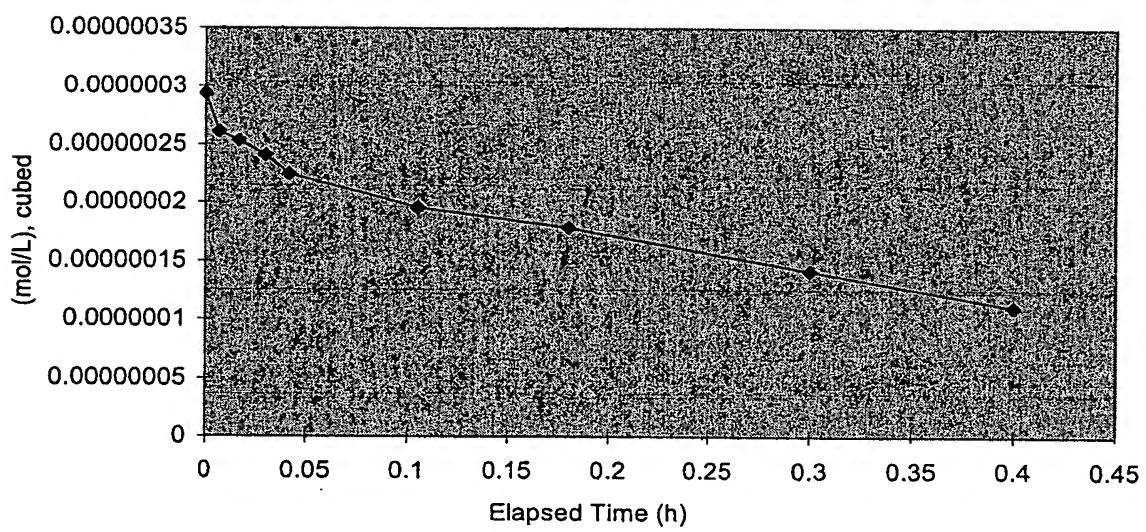


Figure 10: Excess Molar Product (mol/L)³ versus Time (h) Elapsed from pH, OP, and Mg Augmentation

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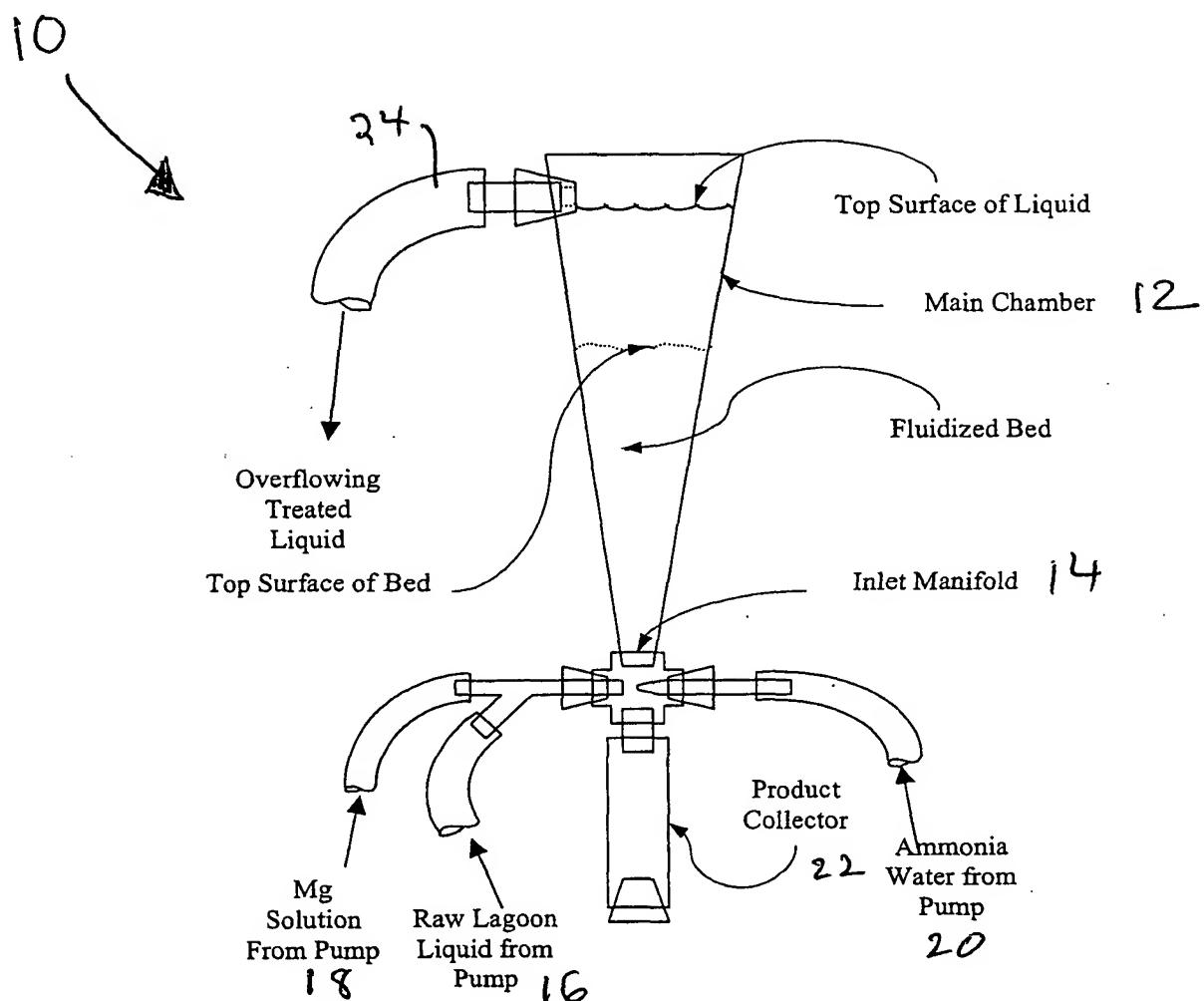
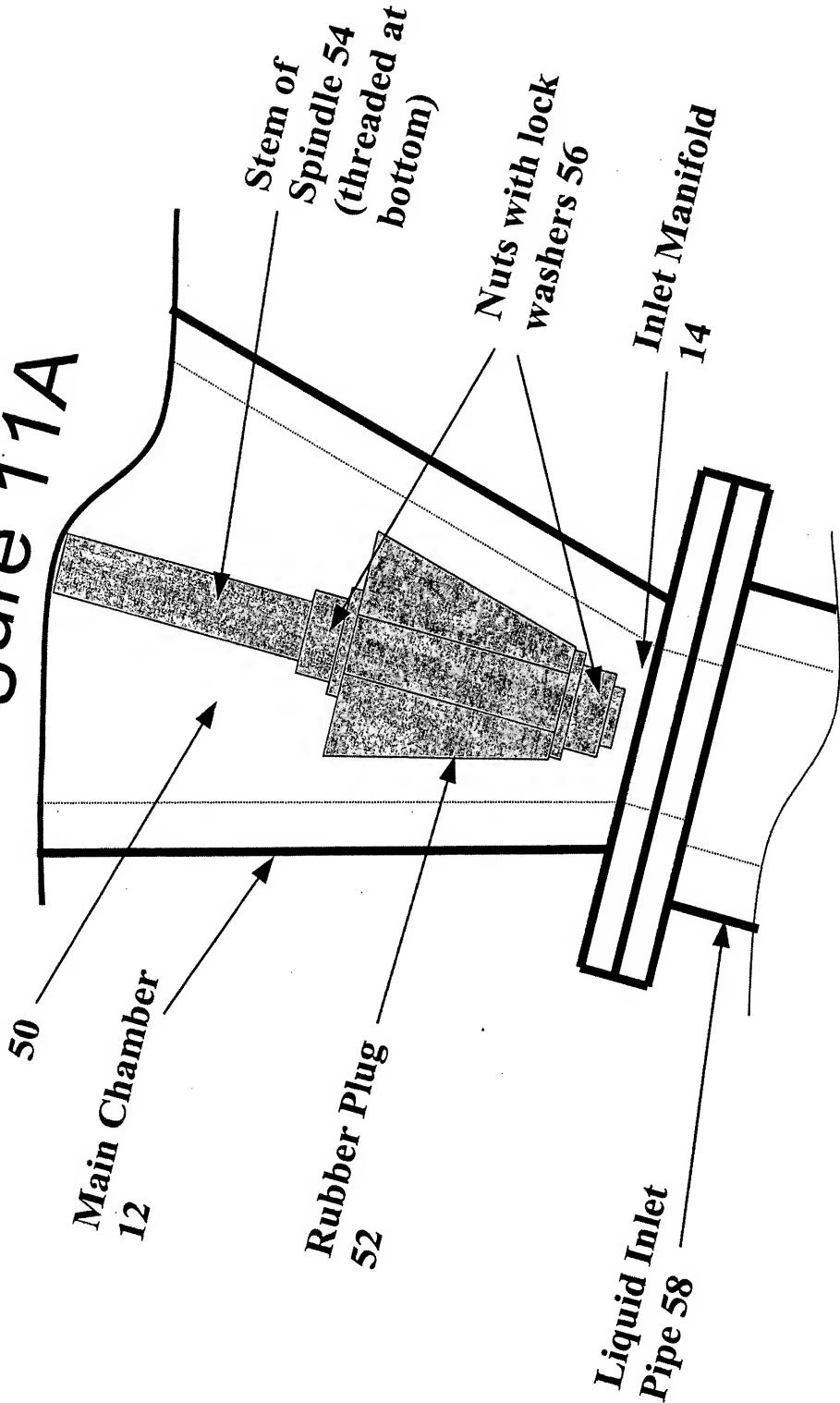


Figure 11 : Sketch of Laboratory-Scale Continuous Crystallizer

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Figure 11A



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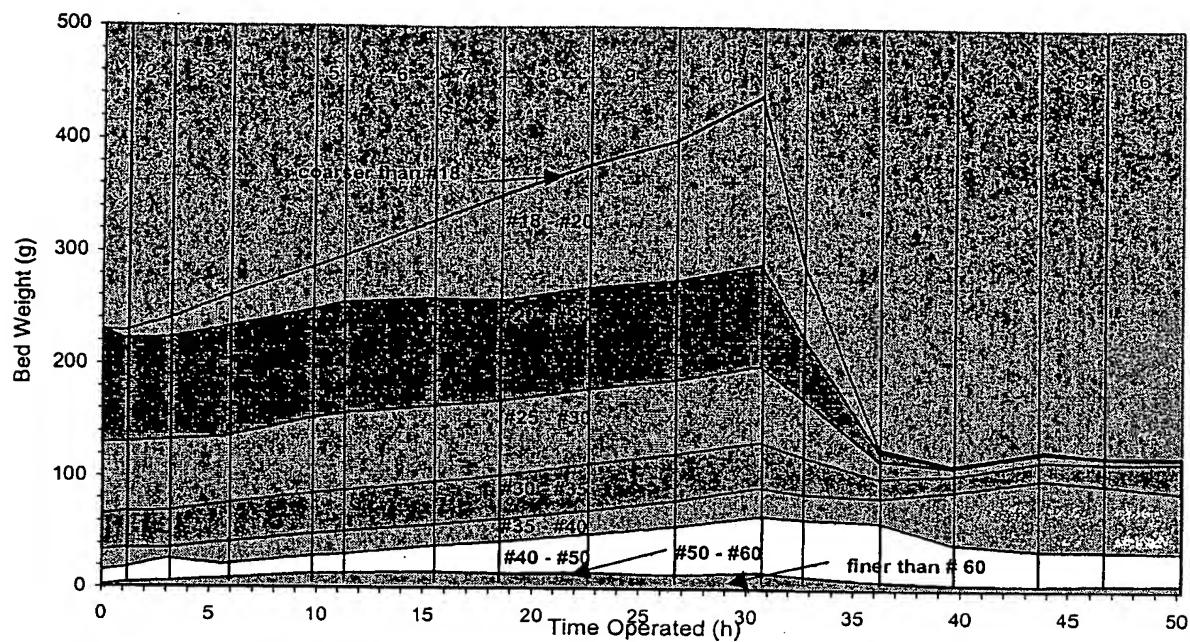


Figure | 2: First Series of FCRs: Bed Weight (g), Broken Down by Particle Size (Standard Sieve), vs. Time Operated (h)
(Numbered Vertical Strips Correspond with Runs)

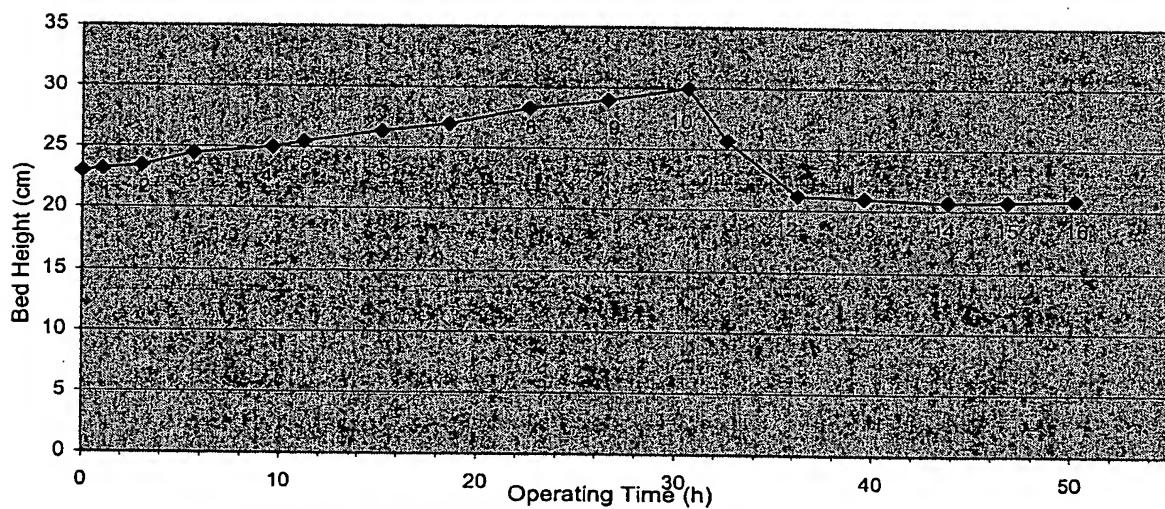
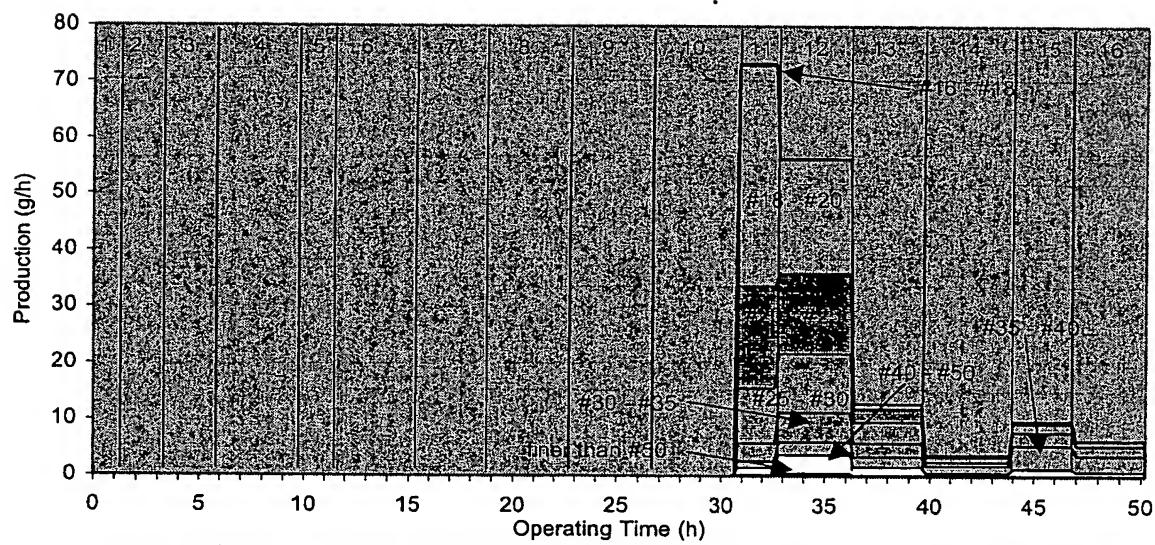


Figure | 3: First Series of FCRs: Bed Height (cm) at End of Run vs. Operating Time (h)
(Run Numbers Indicated)

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**Figure 14: First Series of FCRs: Production (g/h), Averaged Over Each Run,
Broken Down by Particle Size (Standard Sieve)
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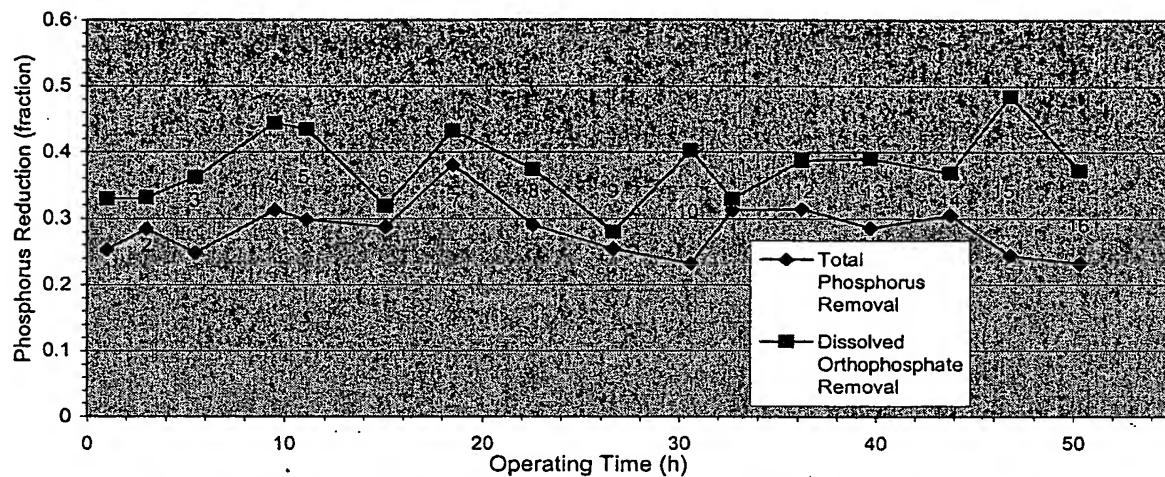


Figure 15: First Series of FCRs: Phosphorus Reduction (fraction) vs.
Operating Time (h)
(Run Numbers Indicated)

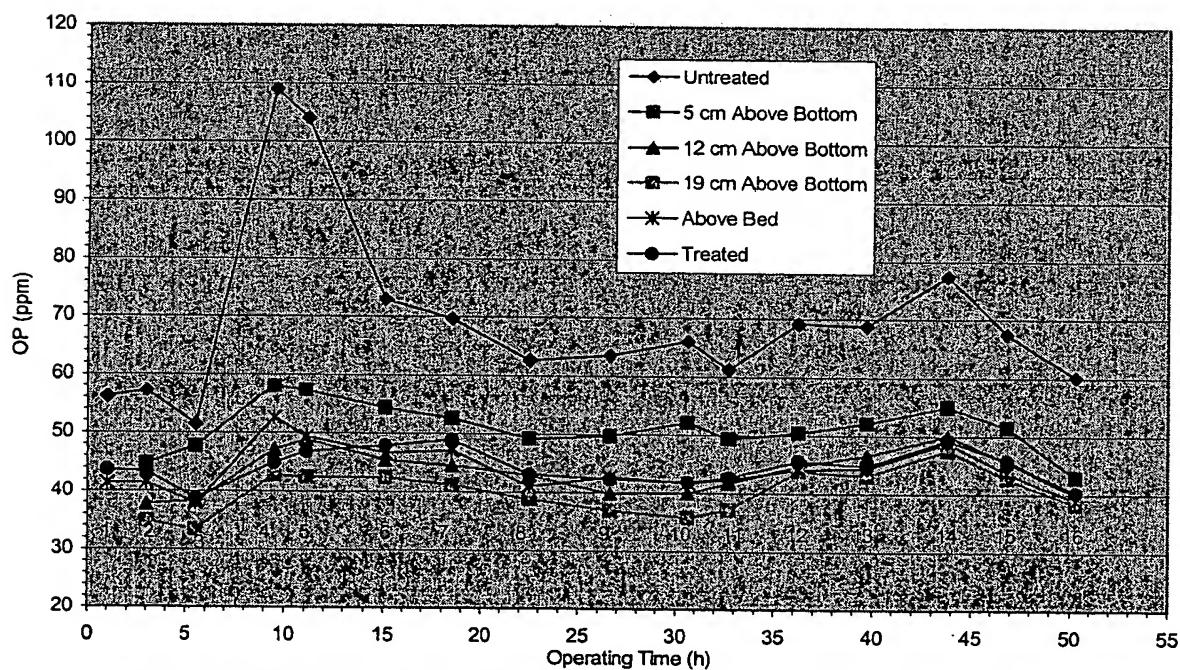


Figure 16: First Series of FCRs: OP (ppm) at Various Sampling Points
vs. Operating Time (h)
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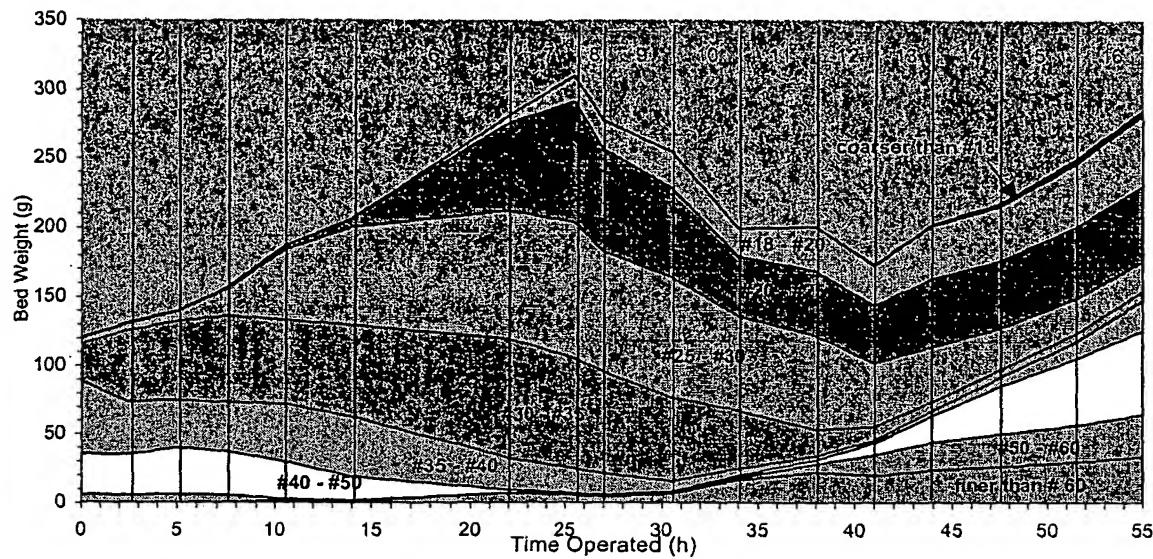


Figure 17: Second Series of FCRs: Bed Weight (g), Broken Down by Particle Size (Standard Sieve), vs. Time Operated (h)
(Numbered Vertical Strips Correspond with Runs)

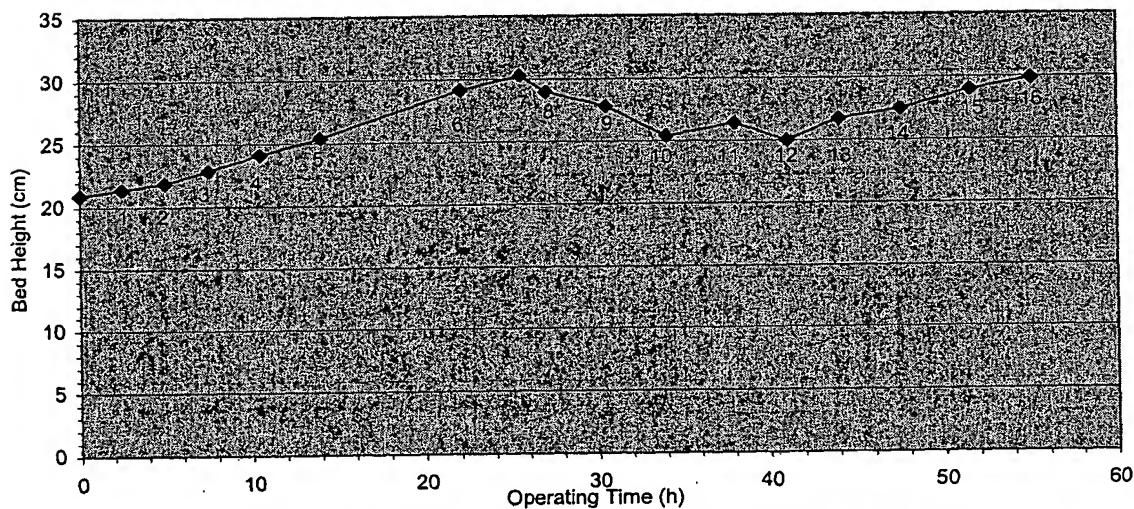


Figure 18: Second Series of FCRs: Bed Height (cm) at End of Run vs.
Operating Time (h)
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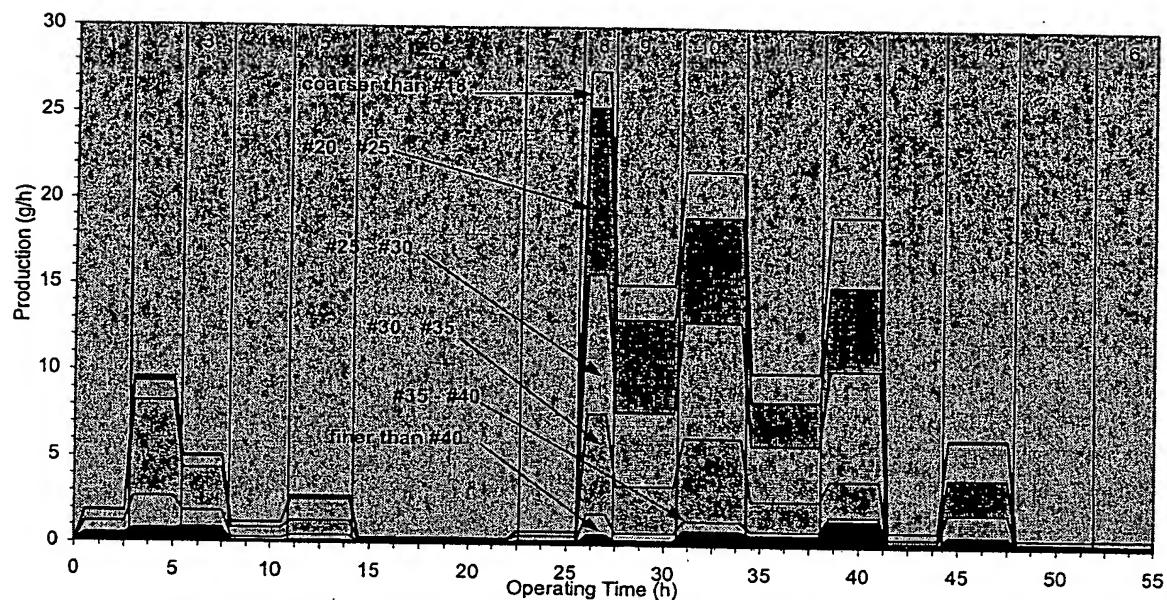


Figure 19: Second Series of FCRs: Production (g/h), Averaged Over Each Run, Broken Down by Particle Size (Standard Sieve)
(Numbered Vertical Strips Correspond with Runs)

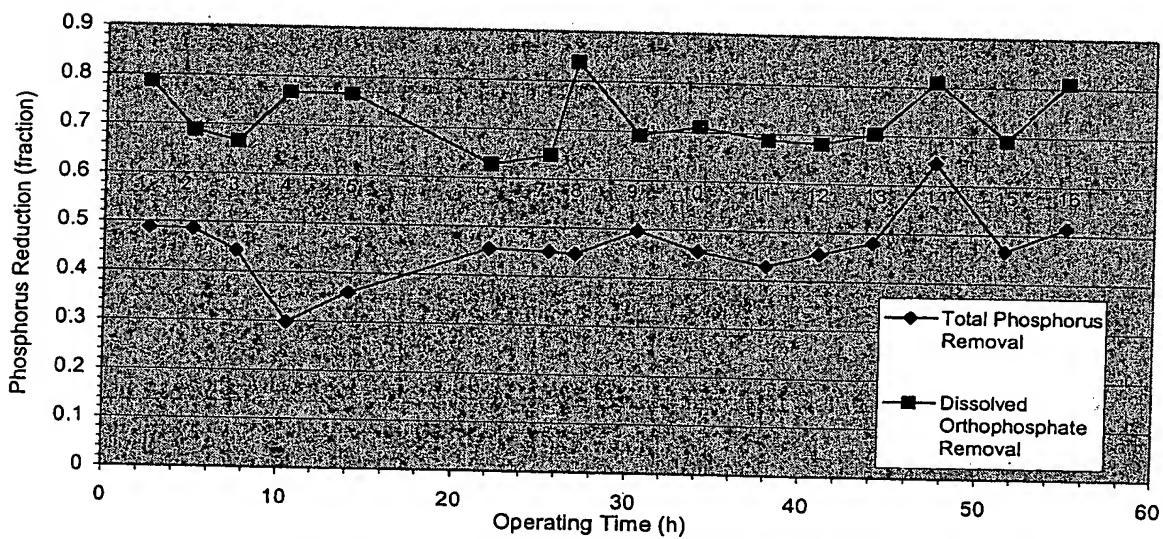


Figure 20: Second Series of FCRs: Phosphorus Reduction (fraction) vs. Operating Time (h)
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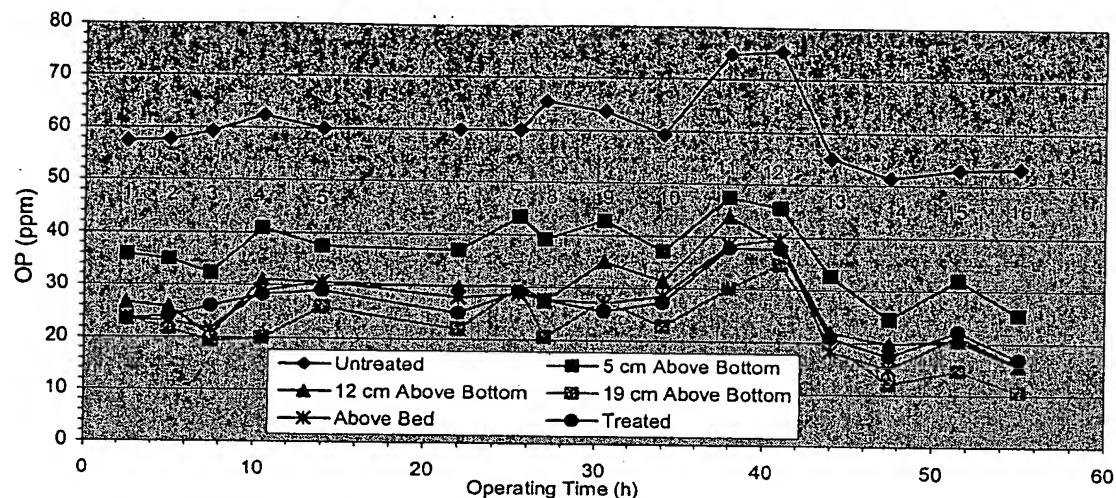


Figure 21: Second Series of FCRs: OP (ppm) at Various Sampling Points
vs. Operating Time (h)
(Run Numbers Indicated)

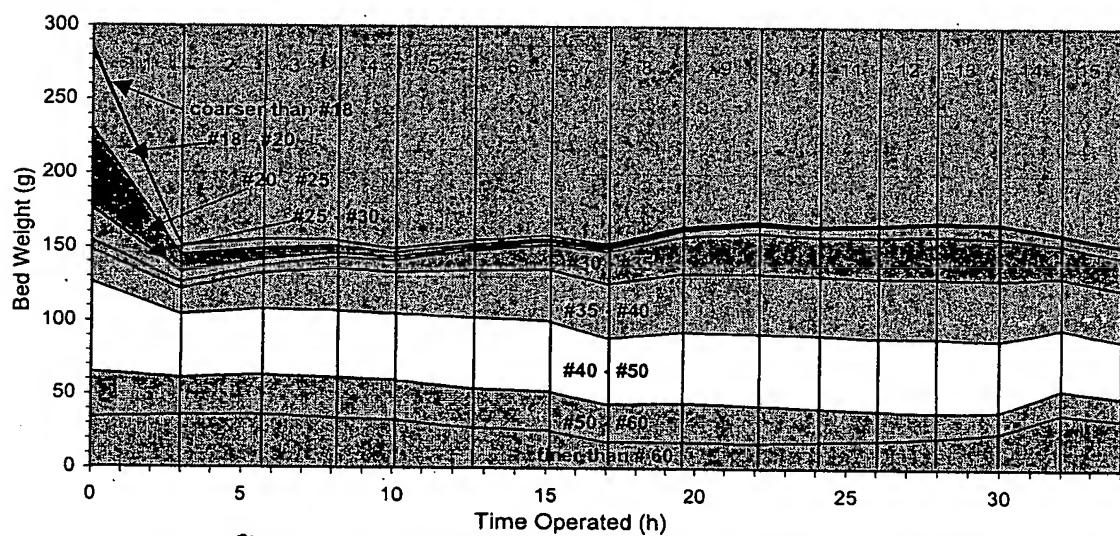


Figure 22: Third Series of FCRs: Bed Weight (g), Broken Down by Particle Size
(Standard Sieve), vs. Time Operated (h)
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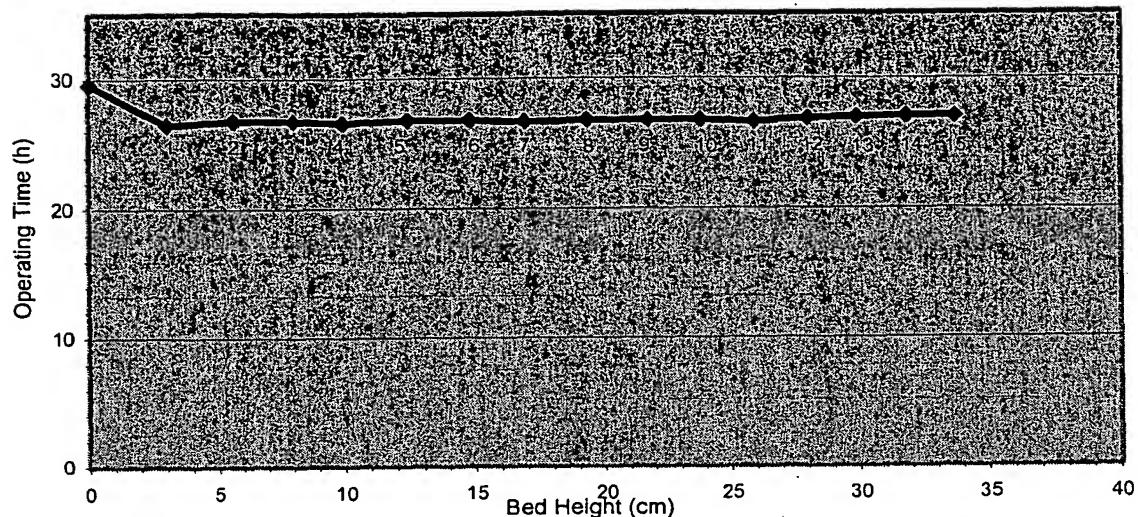


Figure 23: Third Series of FCRs: Bed Height (cm) at End of Run vs.
Operating Time (h)
(Run Numbers Indicated)

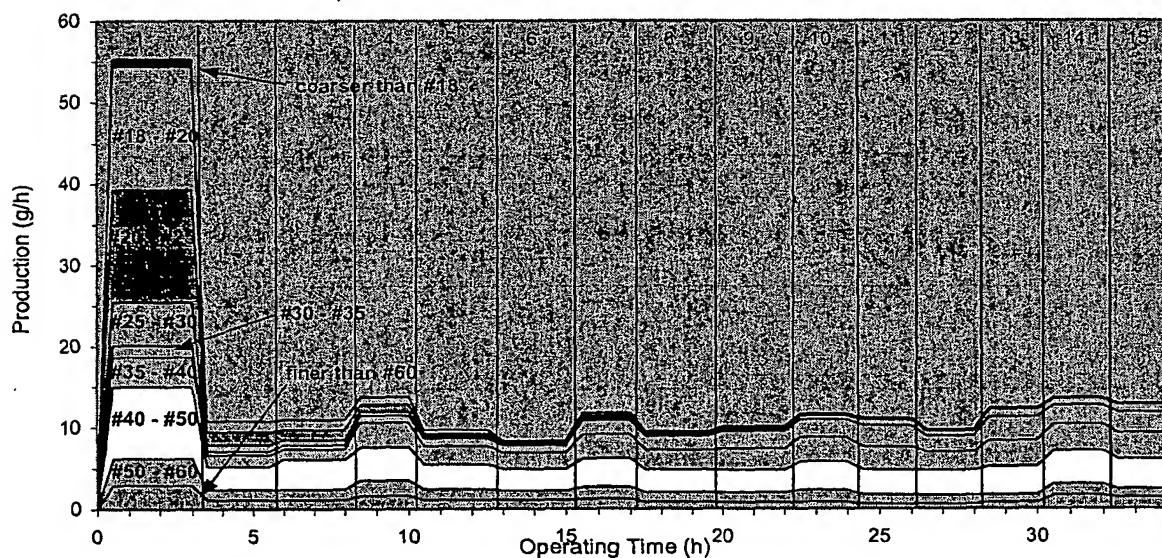


Figure 24: Third Series of FCRs: Production (g/h), Averaged Over Each
Run, Broken Down by Particle Size (Standard Sieve)
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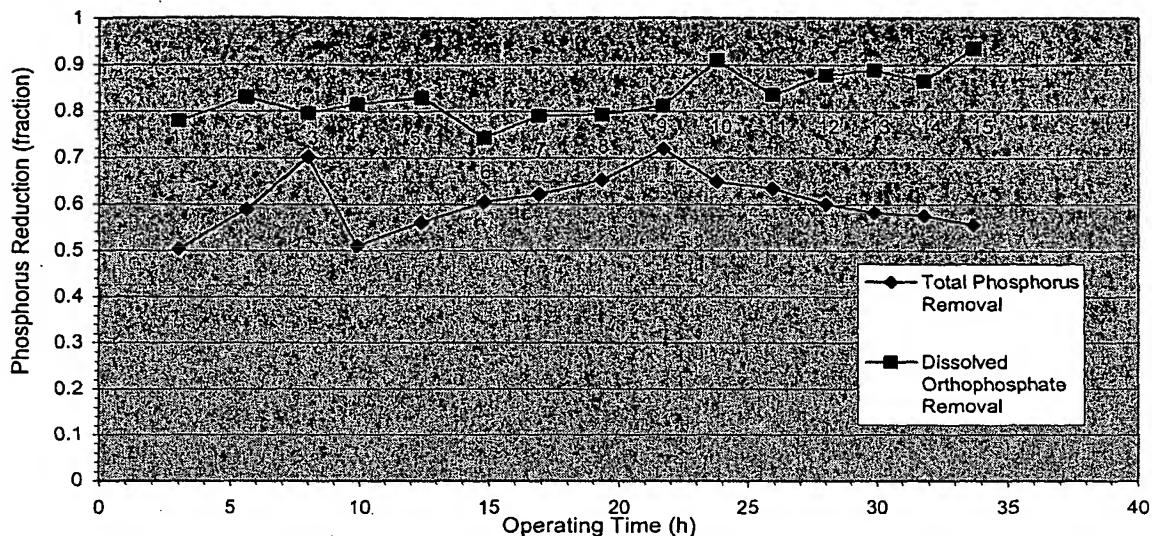


Figure 25: Third Series of FCRs: Phosphorus Reduction (fraction) vs.
Operating Time (h)
(Run Numbers Indicated)

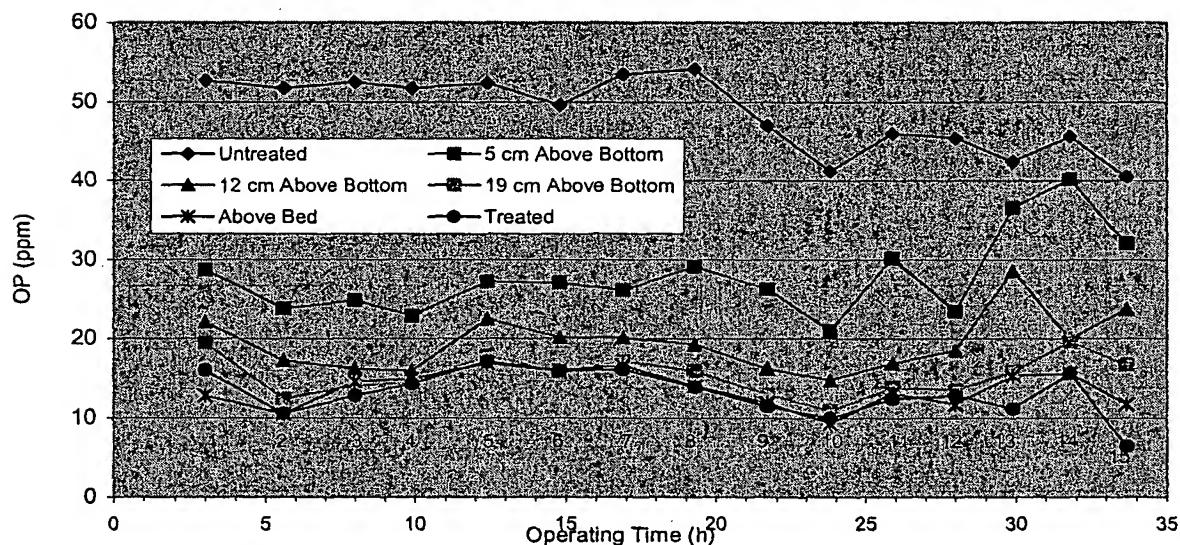


Figure 26: Third Series of FCRs: OP (ppm) at Various Sampling Points
vs. Operating Time (h)
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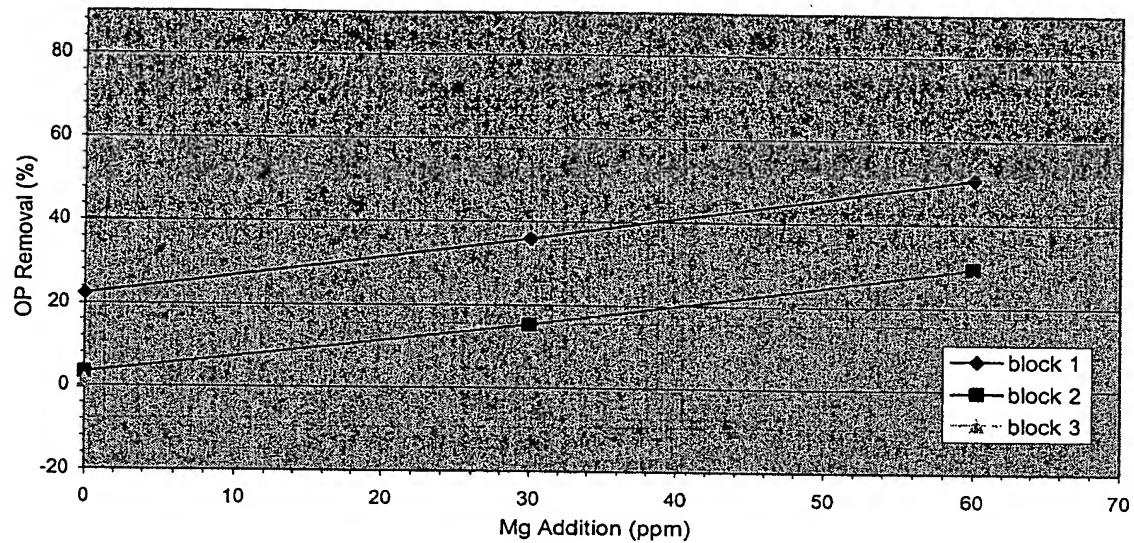


Figure 27: MVRs: OP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 41.2 L/h Flow

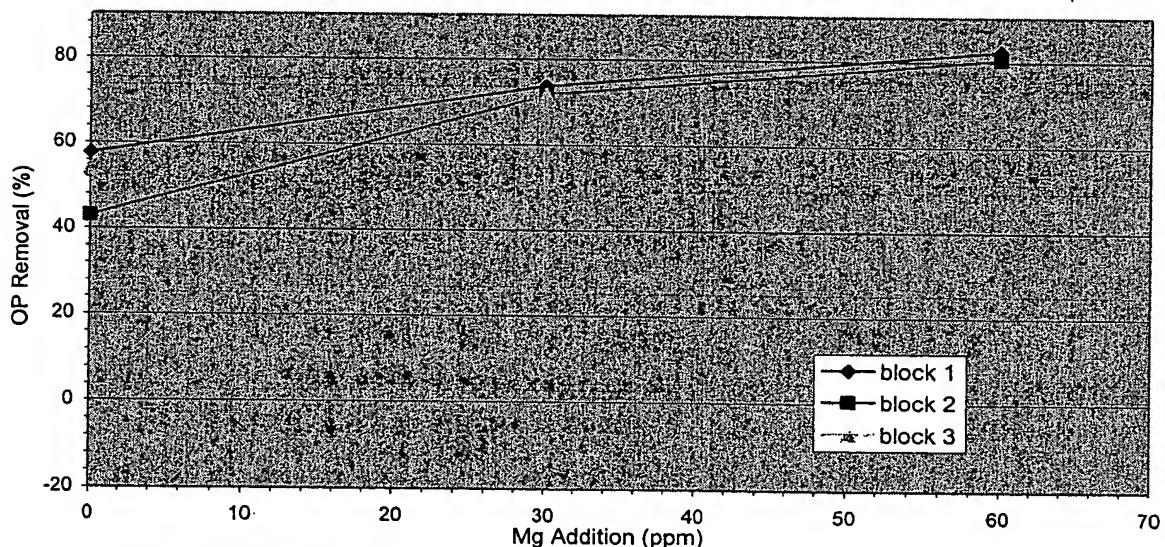


Figure 28: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia Addition and 41.2 L/h Flow

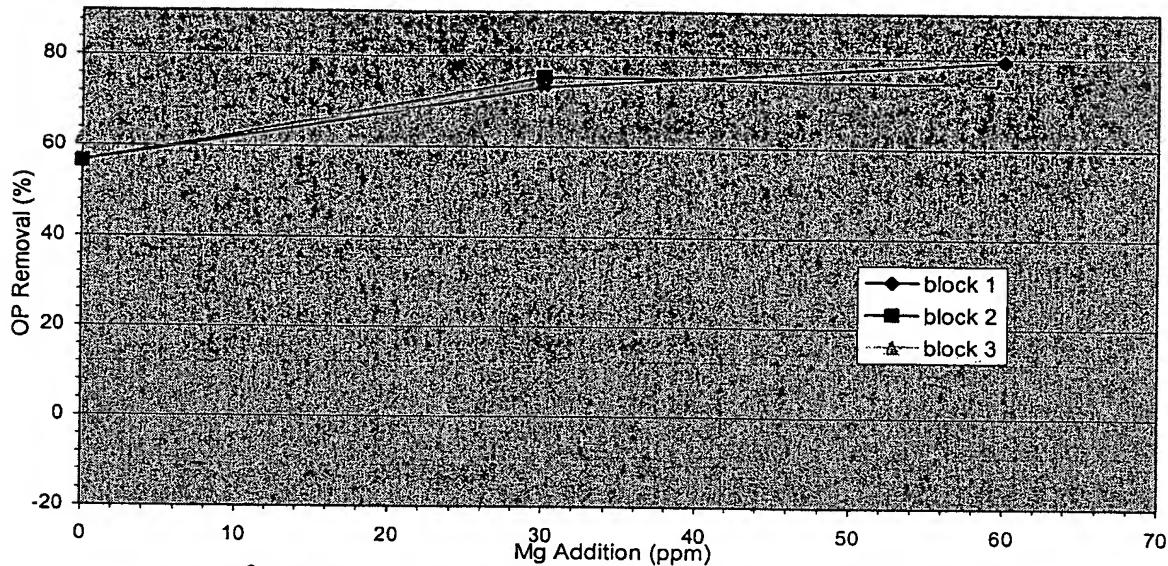


Figure 29: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 200 ppm (as TAN) Ammonia and 41.2 L/h Flow

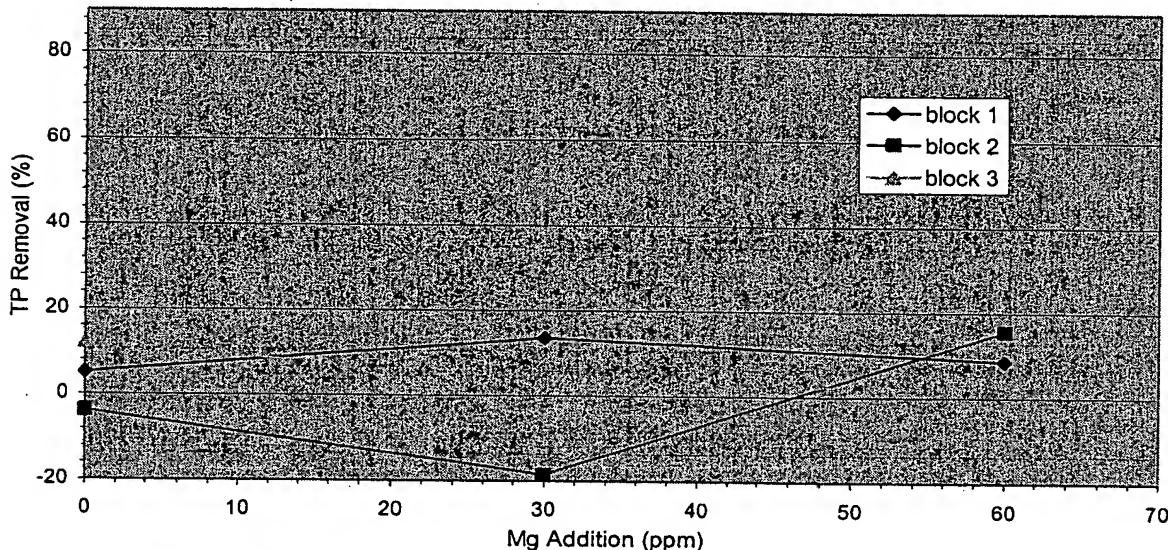


Figure 30: MVRs: TP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 41.2 L/h Flow

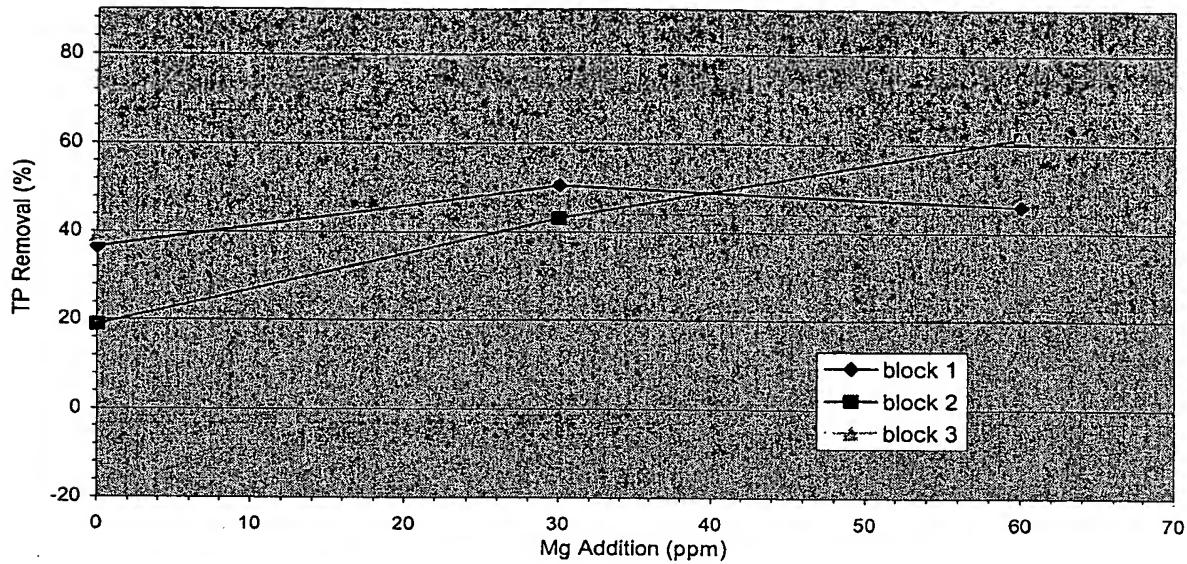


Figure 31: MVRs: TP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia and 41.2 L/h Flow

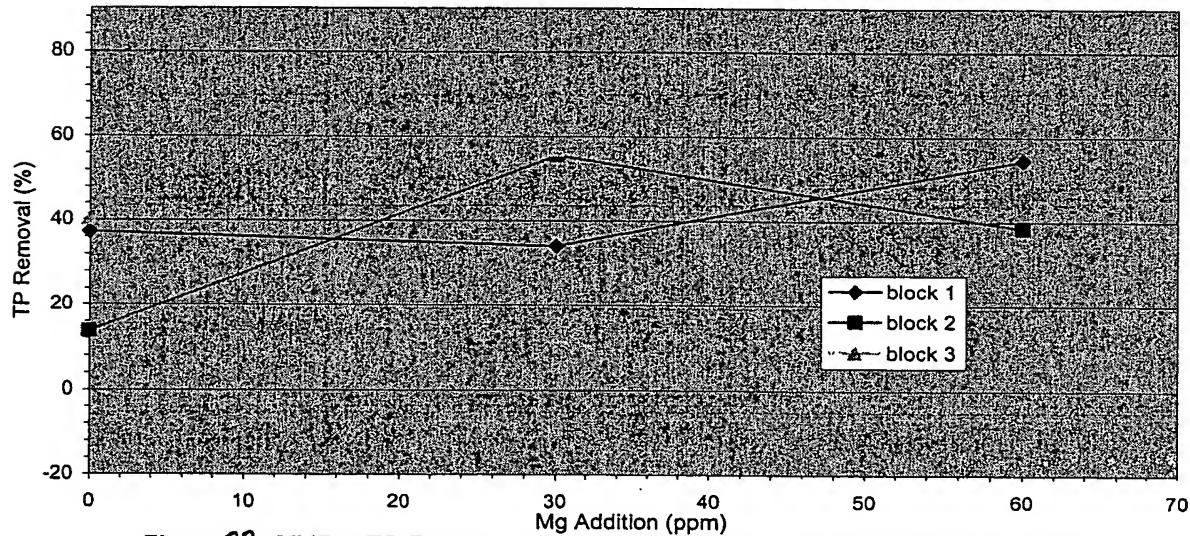


Figure 32: MVRs: TP Removal (%) vs. Mg Addition (ppm) with 200 ppm (as TAN) Ammonia and 41.2 L/h Flow

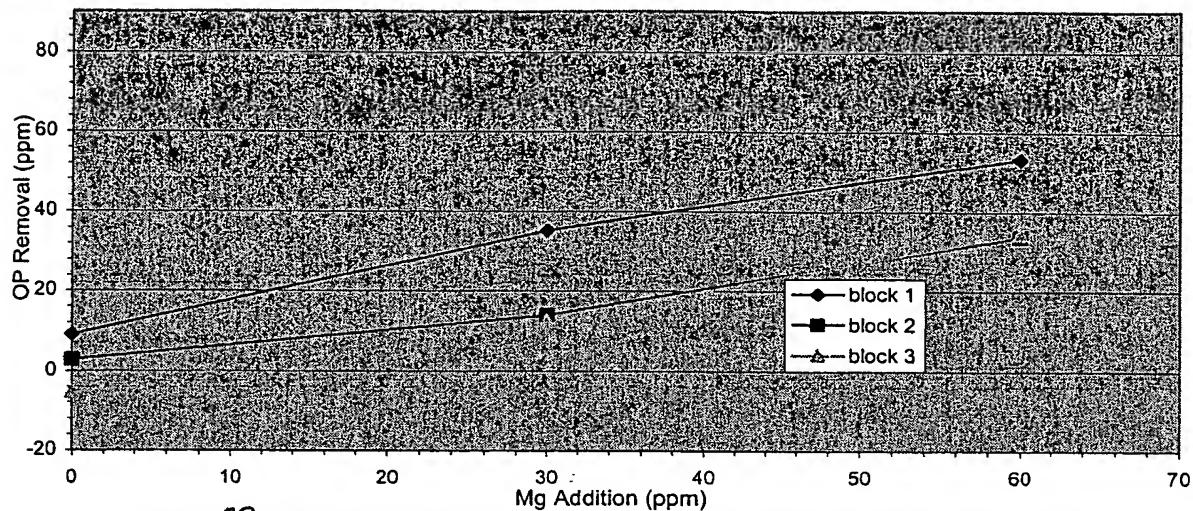


Figure 33: MVRs: OP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 56.8 L/h Flow

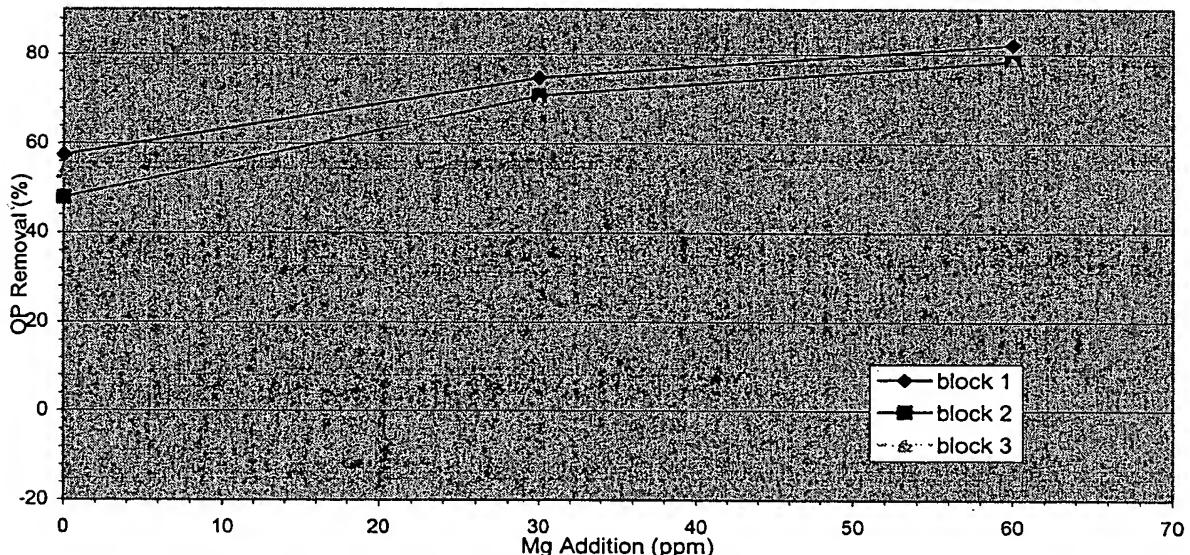


Figure 34: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia and 56.8 L/h Flow

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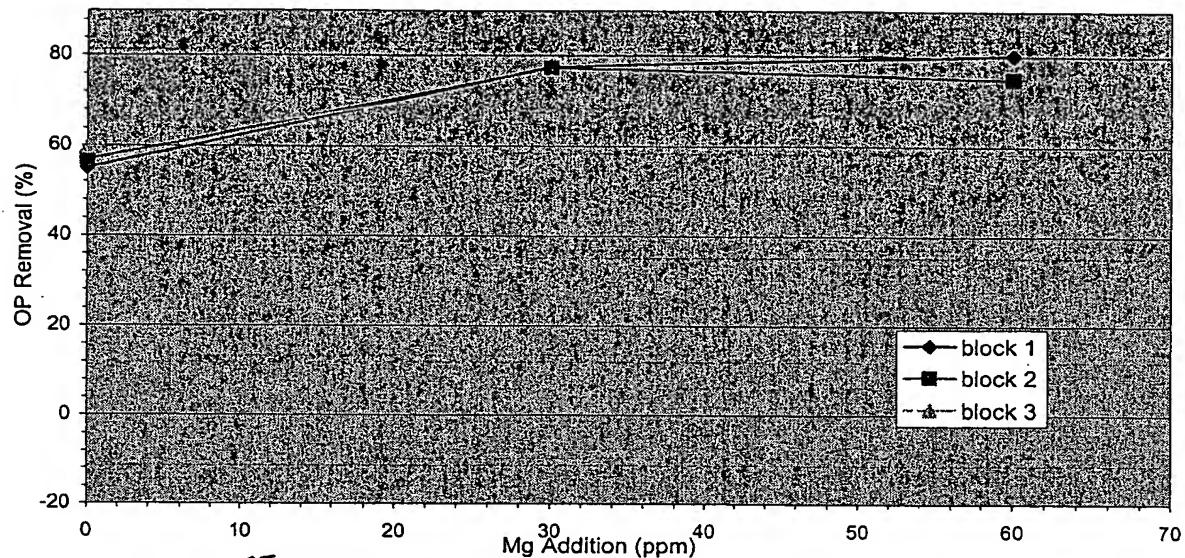


Figure 35: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 200 ppm (as TAN) Ammonia and 56.8 L/h Flow

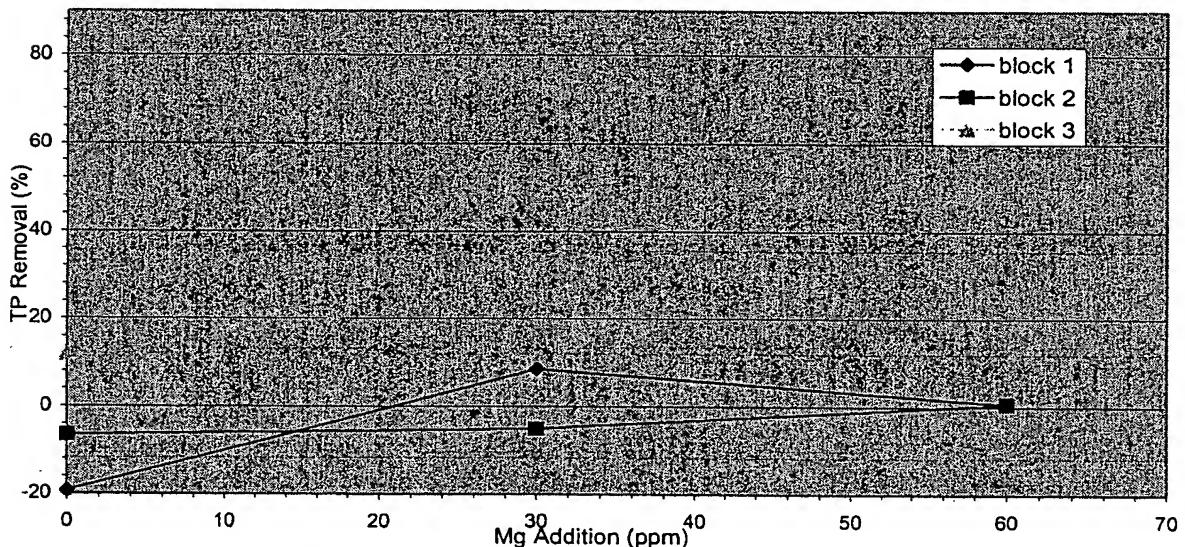


Figure 36: MVRs: TP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 56.8 L/h Flow

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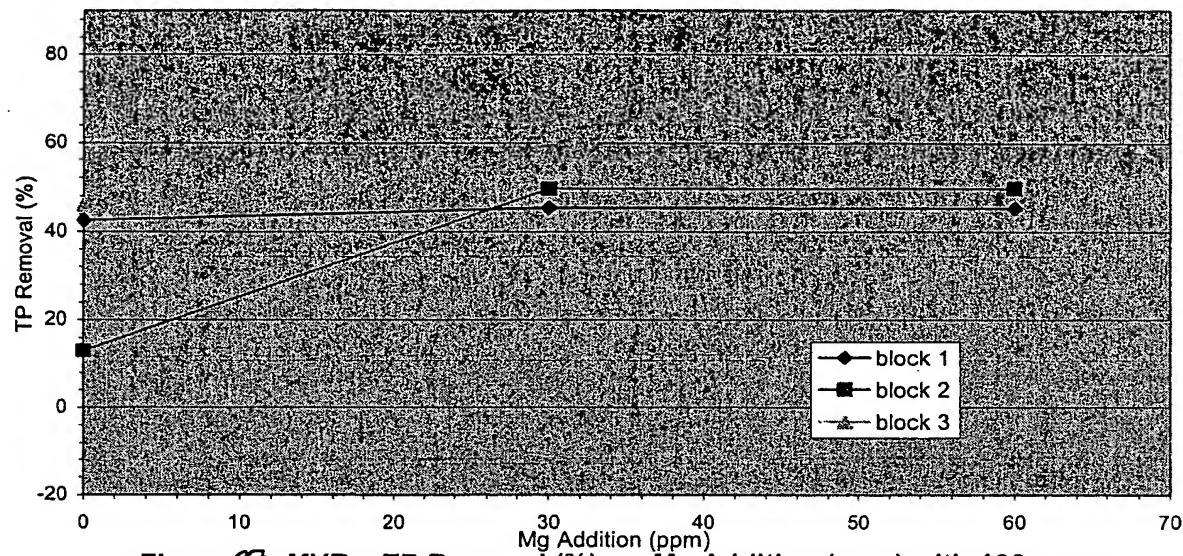


Figure 37: MVRs: TP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as N) Ammonia and 56.8 L/h Flow

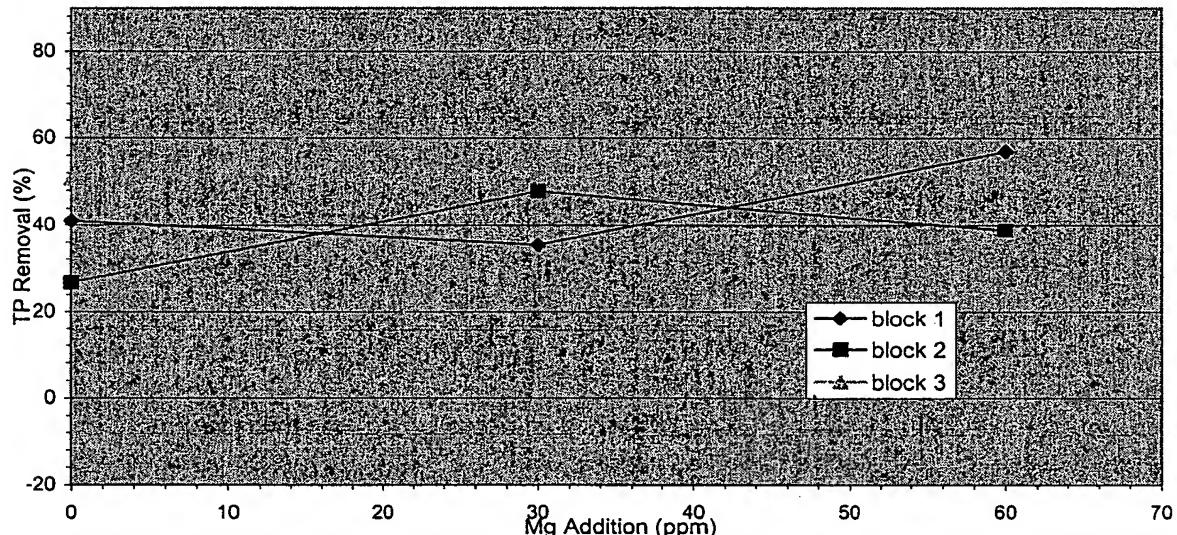


Figure 38: MVRs: TP Removal (%) vs. Mg Addition (ppm) with 200 ppm (as N) Ammonia and 56.8 L/h Flow

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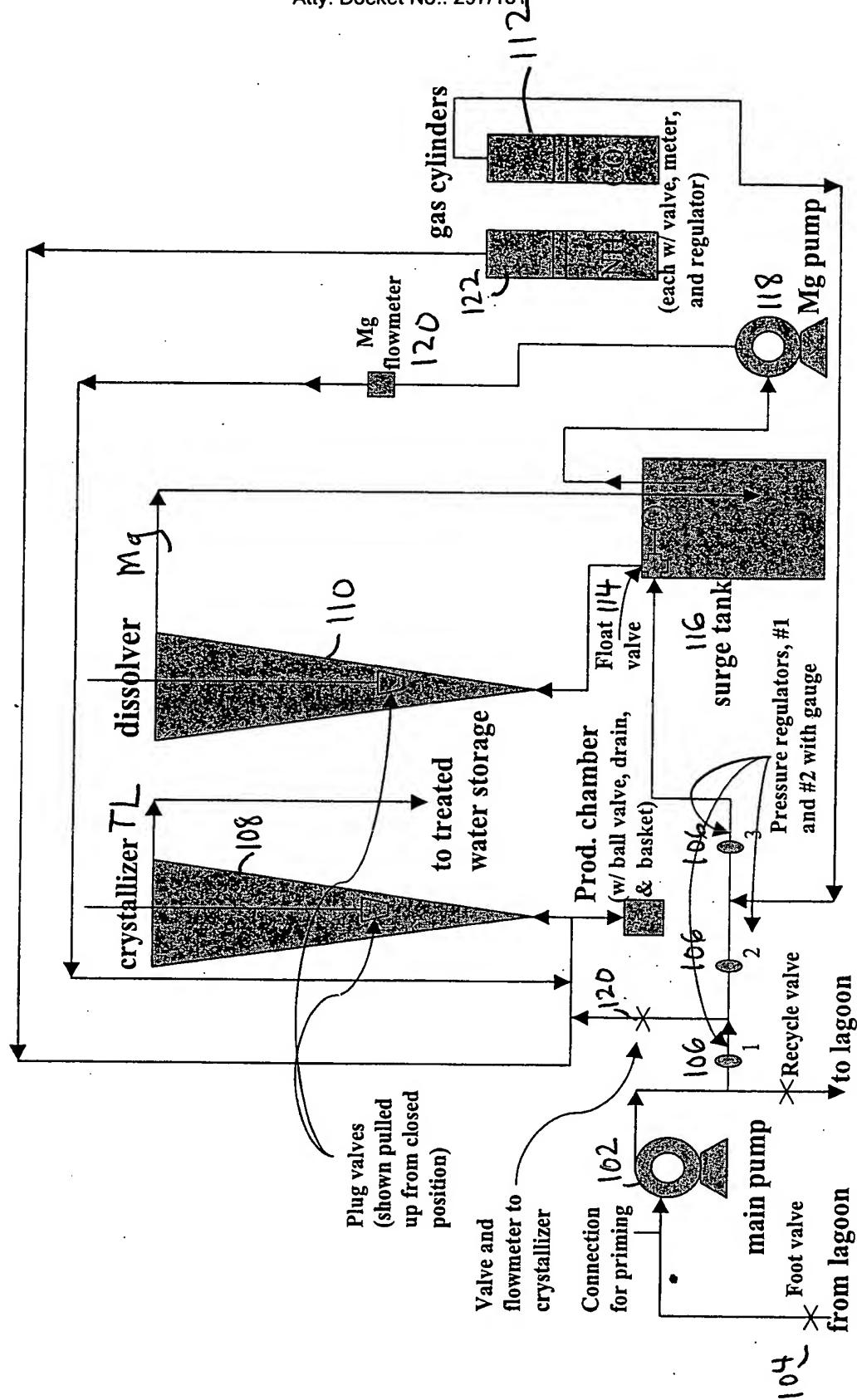


Figure 39: Schematic Representation of Field-Scale Crystallizer, Showing Principal Components

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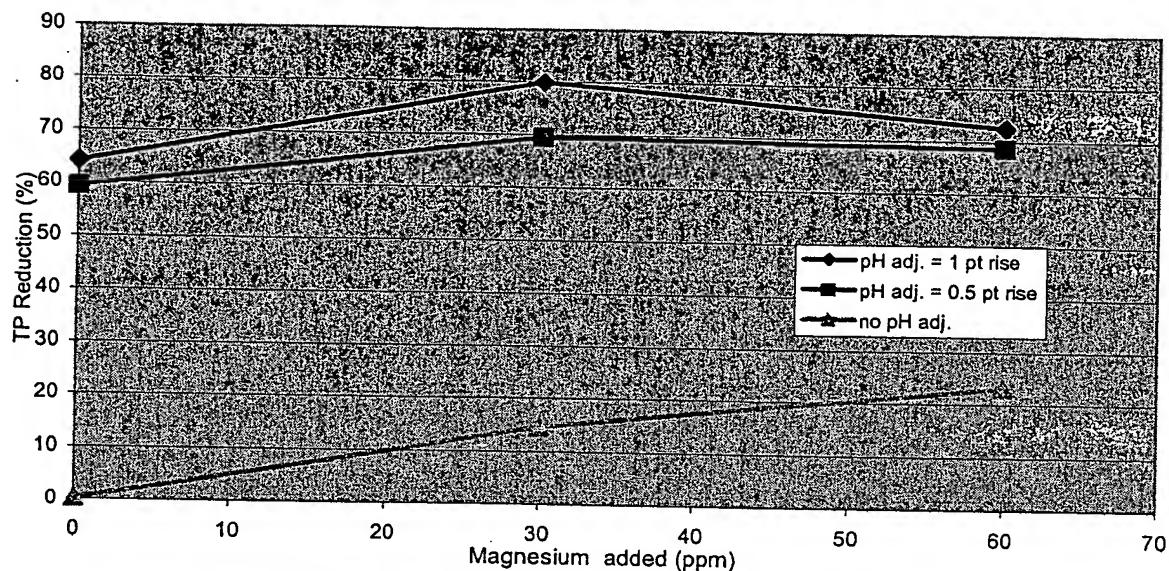


Figure 40: TP Reduction (%) vs. Magnesium added (ppm)
at Lower Flow Rate (341 L/h)

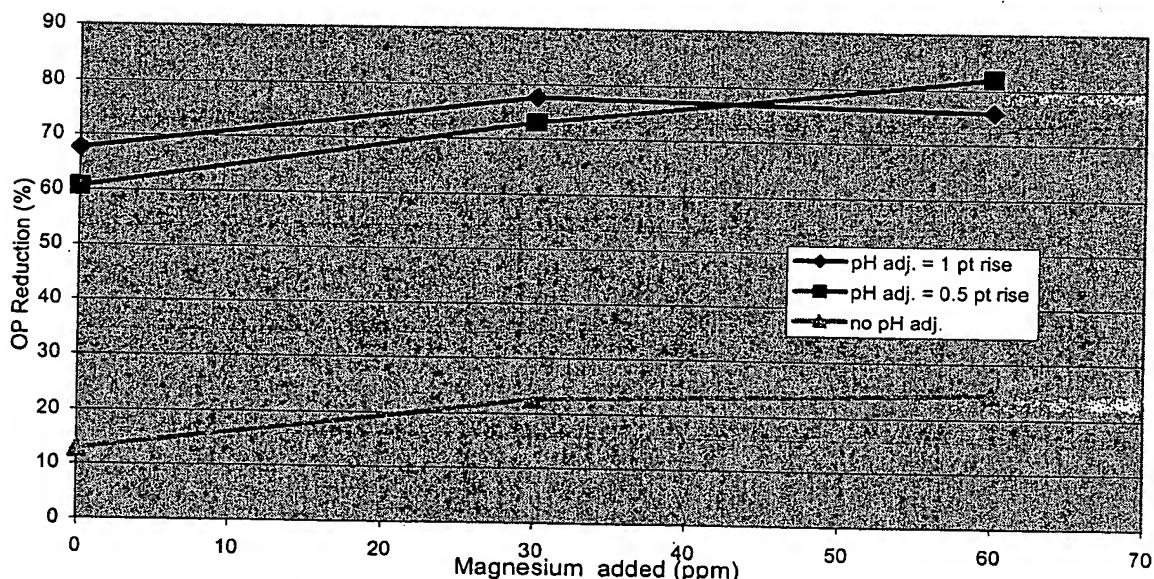


Figure 41: OP Reduction (%) vs. Magnesium added (ppm)
at Lower Flow Rate (341 L/h)

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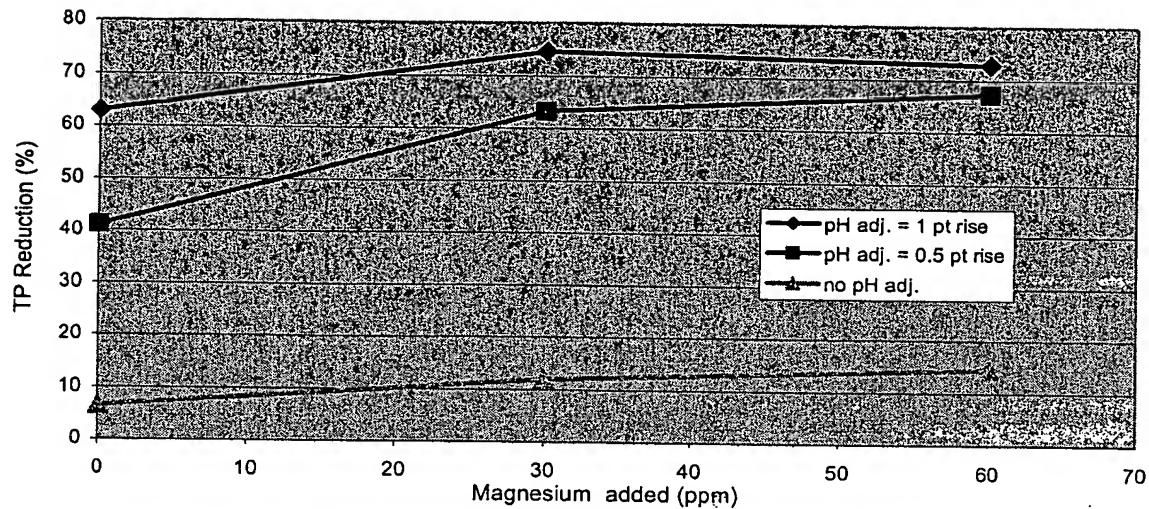


Figure 42: TP Reduction (%) vs. Magnesium added (ppm)
at Higher Flow Rate (568 L/h)

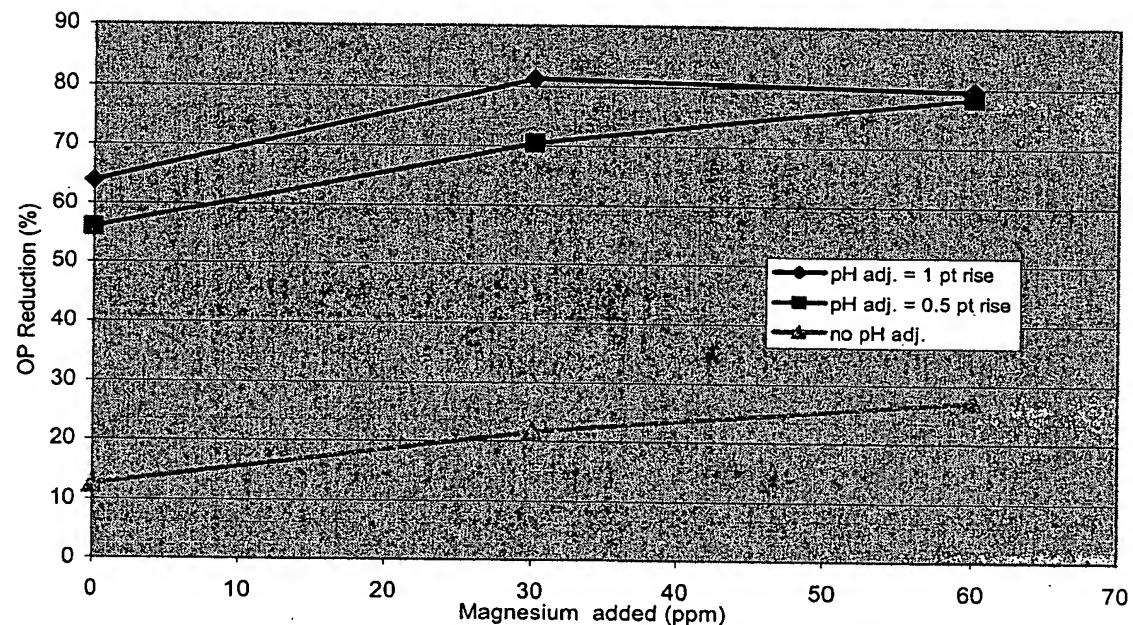


Figure 43: OP Reduction (%) vs. Magnesium added (ppm)
at Higher Flow Rate (568 L/h)